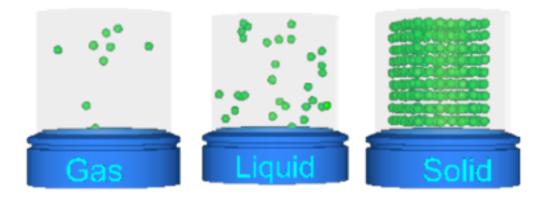


### Entrance Slip

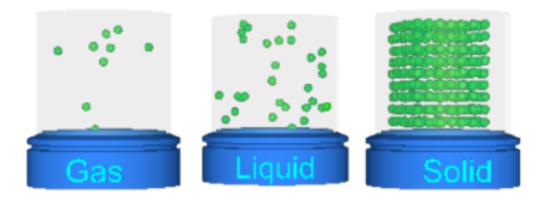
- 1. What is matter?
- 2. Draw and label 2 WHMIS symbols.
- 3. What does MSDS stand for?
- 4. What is the Particle Theory of Matter?





# The Particle Theory of Matter

The <u>particle theory of matter</u> helps to explain what scientists have learned about these tiny particles of matter.





#### Mixtures:

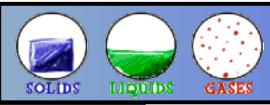
#### Heterogeneous vs. Homogeneous

Heterogeneous mixtures have more than one set of properties.

Homogeneous mixtures have one set of properties.







## Kool-Aid Mixture! Heterogeneous or Homogeneous?



## Kool-Aid Mixture! Heterogeneous or Homogeneous?





### Kool-Aid Mixture! Heterogeneous or Homogeneous?

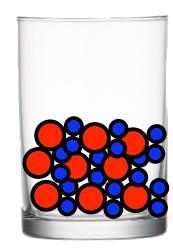
After the Kool-Aid has been mixed together, you can only see one set of properties.







This homogeneous mixture is called a solution.





### Kool-Aid Mixture! Heterogeneous or Homogeneous?

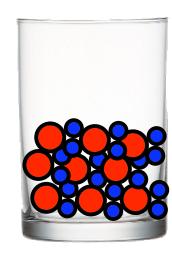
Solutions are mixtures that appear the same all over. The different parts are not visible.







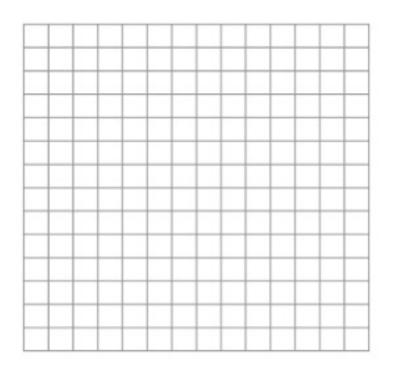
Solutions do not "settle out" - they do not separate by themselves.

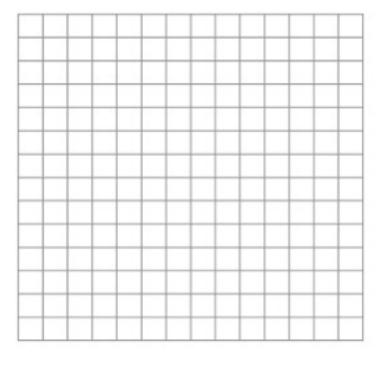




#### Heterogeneous or Homogeneous?

Using graph paper and two pencil crayons, draw a heterogeneous mixture and a homogeneous mixture.

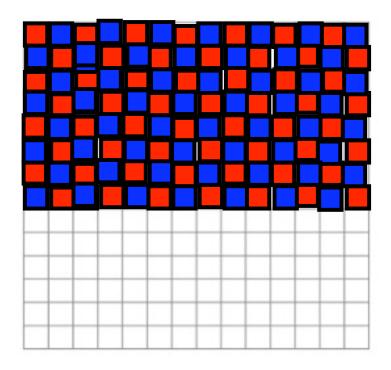


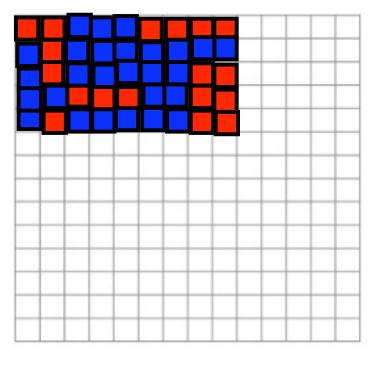




#### Heterogeneous or Homogeneous?

Using graph paper and two pencil crayons, draw a heterogeneous mixture and a homogeneous mixture.







#### **Solutions**



Air -> Nitrogen + Oxygen



Ocean Water -> Salt + Water



#### Mechanical Mixtures

A mechanical mixture is a mixture that has parts that are easily seen by the unaided eye.

Unlike Kool-Aid, we can see the different parts in a mechanical mixture.







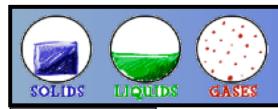
#### Mechanical Mixtures





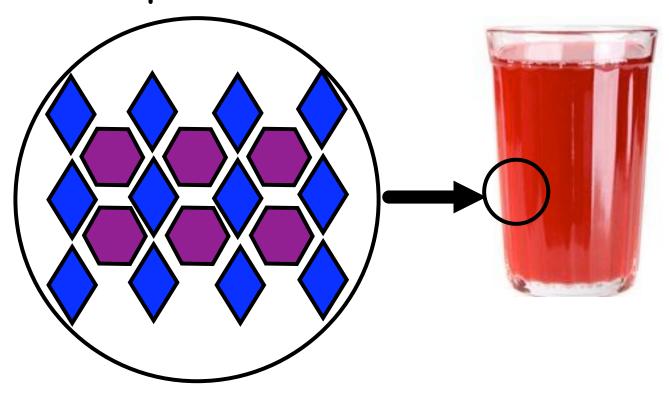


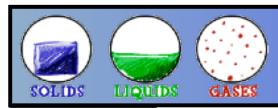




#### Pure Substances

Homogeneous mixtures, like Kool-Aid appear to have one type of particle but we now that it is a mixture of sugar and water particles.





#### Pure Substance

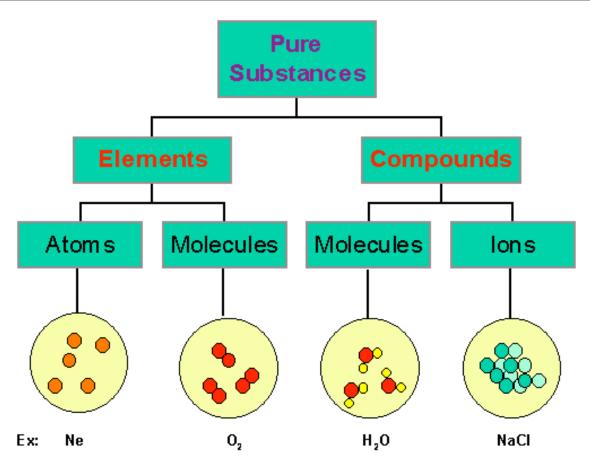
A substance that is only made up of one type of particle is called a pure substance.

A common example is water that has had all of the dissolved minerals removed.









Elements found in the periodic table are pure substances that are made of the exact same particles all over the world.

